

## Cyclopropanation of 3-(2-Oxo-2*H*-chromen-3-yl-carbonyl)-2*H*-chromen-2-one with Zinc Enolates Derived from 1-Aryl-2,2-dibromoalkanones

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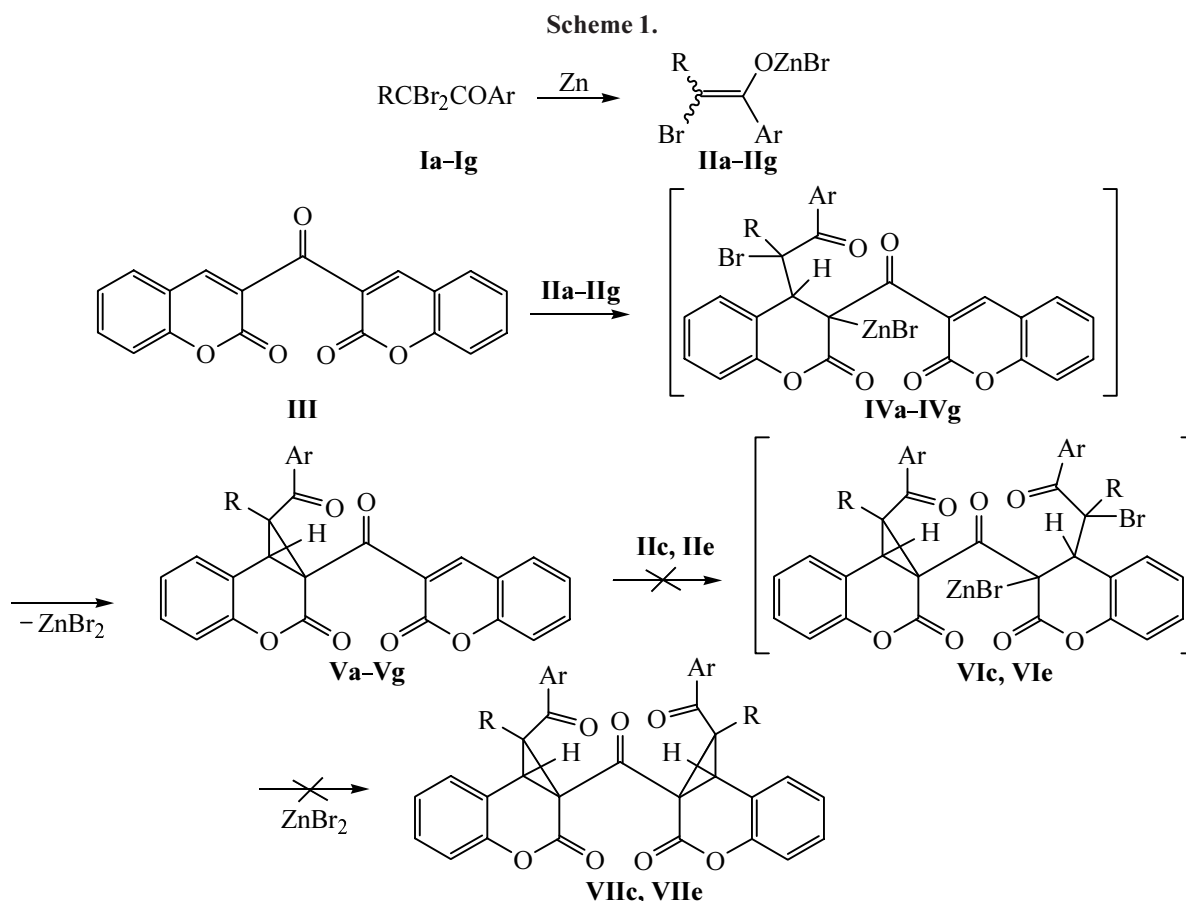
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Received April 21, 2004

**Abstract**—Zinc enolates derived from 1-aryl-2,2-dibromoalkanones react with 3-(2-oxo-2*H*-chromen-3-ylcarbonyl)-2*H*-chromen-2-one to give 1-alkyl-1-aryl-1a-(2-oxo-2*H*-chromen-3-ylcarbonyl)-1a,7b-dihydrocyclopropa[*c*]-chromen-2(1*H*)-ones as a single stereoisomer.

We previously found that bromine-containing zinc enolates react with 3-acetyl-6-bromochromen-2-one to give 1a-acetyl-1-alkyl-1-aryl-6-bromo-1a,7b-dihydrocyclopropa[*c*]chromen-2(1*H*)-ones [1]. In order to further extend the scope of application of this method for

cyclopropanation of substrates possessing an activated double bond, in the present work we examined reactions of 3-(2-oxo-2*H*-chromen-3-ylcarbonyl)-2*H*-chromen-2-one with zinc enolates **IIa–IIg** derived from dibromo ketones **Ia–Ig** (Scheme 1).



R = Me, Ar = 4-MeC<sub>6</sub>H<sub>4</sub> (**a**), 4-FC<sub>6</sub>H<sub>4</sub> (**b**), 4-ClC<sub>6</sub>H<sub>4</sub> (**c**), 4-BrC<sub>6</sub>H<sub>4</sub> (**d**); R = Et, Ar = Ph (**e**), 4-FC<sub>6</sub>H<sub>4</sub> (**f**), 4-ClC<sub>6</sub>H<sub>4</sub> (**g**).

In the first stage, zinc enolates **IIa–IIg** add at one double bond in compound **III** to give intermediates **IVa–IVg** which undergo spontaneous cyclization with formation of fused cyclopropane derivatives, 1-alkyl-1-aryl-1a-(2-oxo-2*H*-chromen-3-ylcarbonyl)-1a,7b-dihydrocyclopropa[*c*]chromen-2(1*H*)-ones **Va–Vg**, in 64–74% yield. We also expected that increase in the zinc enolate-to-substrate ratio (e.g., from 1:1.5 to 1:3) would lead to formation of bis-cyclopropanation products **VII** through intermediates **VI**. However, in the reactions of **III** with excess zinc enolates **IIc** and **IIe** we isolated only monocyclopropane derivatives **Vc** and **Ve**. Presumably, the reason is considerable steric hindrances created by the cyclopropane fragment in **V** to addition of zinc enolate at the activated double bond in the second chromene fragment.

The structure of compounds **Va–Vg** was confirmed by the data of elemental analysis and IR and <sup>1</sup>H NMR spectroscopy. The IR spectra of **Va–Vg** contain characteristic carbonyl absorption bands at 1670–1680 (COAr, COC=C) and 1725–1740 cm<sup>-1</sup> (COO). In the <sup>1</sup>H NMR spectra of these compounds, apart from other signals, we observed singlets at δ 3.69–3.73 and 8.10–8.18 ppm, which belong to the 7b-H and 4'-H protons, respectively. The presence of only one set of signals for each compound **V** indicates formation of a single stereoisomer (with respect to the cyclopropane ring plane).

## EXPERIMENTAL

The IR spectra were recorded on a UR-20 spectrometer from samples dispersed in mineral oil. The <sup>1</sup>H NMR spectra were recorded on a Tesla BS-576A spectrometer (80 MHz) from solutions in CDCl<sub>3</sub>; the chemical shifts were measured relative to hexamethyl-disiloxane as internal reference.

**1-Alkyl-1-aryl-1a-(2-oxo-2*H*-chromen-3-ylcarbonyl)-1a,7b-dihydrocyclopropa[*c*]chromen-2(1*H*)-ones Va–Vg (general procedure).** A solution of 0.0075 mol of dibromo ketone **Ia–Ig** in 3 ml of ethyl acetate was added to a mixture of 2 g of zinc (prepared as fine turnings), 8 ml of diethyl ether, and 5 ml of ethyl acetate. The mixture was heated to initiate the reaction, and it then occurred spontaneously. When the reaction was complete, the mixture was heated for 15 min on a water bath and cooled, and the liquid phase was separated from unreacted zinc by decanting into another flask. Compound **III**, 0.005 mol, and 25 ml of ethyl acetate were added, and the mixture was heated for 30 min on

a water bath, cooled, treated with 5% hydrochloric acid, and extracted with diethyl ether. The extract was dried over sodium sulfate and evaporated, and the residue was recrystallized thrice from a mixture of ethyl acetate, carbon tetrachloride, and methanol at a ratio of 1:1:3. In the synthesis of compounds **Vc** and **Ve**, 0.015 mol of dibromo ketone **IIc** or **IIe** was used.

**1-Methyl-1-(4-methylbenzoyl)-1a-(2-oxo-2*H*-chromen-3-ylcarbonyl)-1a,7b-dihydrocyclopropa[*c*]chromen-2(1*H*)-one (Va).** Yield 74%, mp 149–150°C. <sup>1</sup>H NMR spectrum, δ, ppm: 1.22 s (3H, Me), 2.33 s (3H, 4-MeC<sub>6</sub>H<sub>4</sub>), 3.73 s (1H, CH), 6.90–7.80 m (12H, C<sub>6</sub>H<sub>4</sub>, MeC<sub>6</sub>H<sub>4</sub>), 8.18 s (1H, CH=). Found, %: C 74.85; H 4.26. C<sub>29</sub>H<sub>20</sub>O<sub>6</sub>. Calculated, %: C 74.99; H 4.34.

**1-(4-Fluorobenzoyl)-1-methyl-1a-(2-oxo-2*H*-chromen-3-ylcarbonyl)-1a,7b-dihydrocyclopropa[*c*]chromen-2(1*H*)-one (Vb).** Yield 65%, mp 155–157°C. <sup>1</sup>H NMR spectrum, δ, ppm: 1.21 s (3H, Me), 3.71 s (1H, CH), 6.90–7.90 m (12H, C<sub>6</sub>H<sub>4</sub>, 4-FC<sub>6</sub>H<sub>4</sub>), 8.15 s (1H, CH=). Found, %: C 71.92; H 3.58. C<sub>28</sub>H<sub>17</sub>FO<sub>6</sub>. Calculated, %: C 71.79; H 3.66.

**1-(4-Chlorobenzoyl)-1-methyl-1a-(2-oxo-2*H*-chromen-3-ylcarbonyl)-1a,7b-dihydrocyclopropa[*c*]chromen-2(1*H*)-one (Vc).** Yield 67%, mp 160–162°C. <sup>1</sup>H NMR spectrum, δ, ppm: 1.21 s (3H, Me), 3.70 s (1H, CH), 6.90–7.90 m (12H, C<sub>6</sub>H<sub>4</sub>, 4-ClC<sub>6</sub>H<sub>4</sub>), 8.16 s (1H, CH=). Found, %: C 69.22; H 3.46. C<sub>28</sub>H<sub>17</sub>ClO<sub>6</sub>. Calculated, %: C 69.36; H 3.53.

**1-(4-Bromobenzoyl)-1-methyl-1a-(2-oxo-2*H*-chromen-3-ylcarbonyl)-1a,7b-dihydrocyclopropa[*c*]chromen-2(1*H*)-one (Vd).** Yield 71%, mp 162–163°C. <sup>1</sup>H NMR spectrum, δ, ppm: 1.21 s (3H, Me), 3.71 s (1H, CH), 6.90–7.80 m (12H, C<sub>6</sub>H<sub>4</sub>, 4-BrC<sub>6</sub>H<sub>4</sub>), 8.17 s (1H, CH=). Found, %: C 63.41; H 3.30. C<sub>28</sub>H<sub>17</sub>BrO<sub>6</sub>. Calculated, %: C 63.53; H 3.24.

**1-Benzoyl-1-ethyl-1a-(2-oxo-2*H*-chromen-3-ylcarbonyl)-1a,7b-dihydrocyclopropa[*c*]chromen-2(1*H*)-one (Ve).** Yield 64%, mp 134–136°C. <sup>1</sup>H NMR spectrum, δ, ppm: 0.43 t (3H, CH<sub>2</sub>CH<sub>3</sub>, *J* = 7 Hz), 1.30 m, 2.10 m (2H, CH<sub>2</sub>CH<sub>3</sub>), 3.71 s (1H, CH), 6.90–7.90 m (13H, C<sub>6</sub>H<sub>4</sub>, C<sub>6</sub>H<sub>5</sub>), 8.10 s (1H, CH=). Found, %: C 74.85; H 4.39. C<sub>29</sub>H<sub>20</sub>O<sub>6</sub>. Calculated, %: C 74.99; H 4.34.

**1-Ethyl-1-(4-fluorobenzoyl)-1a-(2-oxo-2*H*-chromen-3-ylcarbonyl)-1a,7b-dihydrocyclopropa[*c*]chromen-2(1*H*)-one (Vf).** Yield 65%, mp 126–127°C. <sup>1</sup>H NMR spectrum, δ, ppm: 0.45 t (3H, CH<sub>2</sub>CH<sub>3</sub>,

$J = 7$  Hz), 1.30 m, 2.10 m (2H,  $\text{CH}_2\text{CH}_3$ ), 3.69 s (1H, CH), 6.90–7.95 m (12H,  $\text{C}_6\text{H}_4$ , 4- $\text{FC}_6\text{H}_4$ ), 8.10 s (1H, CH=). Found, %: C 72.01; H 3.87.  $\text{C}_{29}\text{H}_{19}\text{FO}_6$ . Calculated, %: C 72.20; H 3.97

**1-(4-Chlorobenzoyl)-1-ethyl-1a-(2-oxo-2H-chromen-3-ylcarbonyl)-1a,7b-dihydrocyclopropa-[c]chromen-2(1H)-one (Vg).** Yield 70%, mp 174–175°C.  $^1\text{H}$ NMR spectrum,  $\delta$ , ppm: 0.45 t (3H,  $\text{CH}_2\text{CH}_3$ ,  $J = 7$  Hz), 1.30 m, 2.10 m (2H,  $\text{CH}_2\text{CH}_3$ ), 3.69 s (1H, CH), 6.90–7.90 m (12H,  $\text{C}_6\text{H}_4$ , 4- $\text{ClC}_6\text{H}_4$ ), 8.10 s (1H,

CH=). Found, %: C 69.70; H 3.71.  $\text{C}_{29}\text{H}_{19}\text{ClO}_6$ . Calculated, %: C 69.81; H 3.84.

This study was performed under financial support by the Russian Foundation for Basic Research (project no. 04-03-96036).

#### REFERENCE

1. Shchepin, V.V., Russkikh, N.Yu., Kalyuzhnyi, M.M., Shchepin, R.V., and Vakhrin, M.I., *Russ. J. Org. Chem.*, 2003, vol. 39, p. 1316.